## (Supplementary Section)

## Facile synthesis of ZIF-67 incorporated electrospun PVA nanofibers composite for

## efficient Pb (II) adsorption from water: Docking and experimental studies

Simranjeet Singh<sup>1</sup>, Pavithra N<sup>1</sup>, Basavaraju Uppara<sup>2,3</sup>, Radhika Varshney<sup>1</sup>, Nabila Shehata<sup>4</sup>, Nadeem A Khan<sup>5</sup>, Jinu Joji<sup>3</sup>, Joginder Singh<sup>6</sup>, Praveen C Ramamurthy<sup>1\*</sup>

<sup>1</sup>Interdisciplinary Centre for Water Research (ICWaR), Indian Institute of Science, Bengaluru, Karnataka, India

<sup>2</sup>Centre for Smart Manufacturing Precision Machine Tools and Aggregates, Central Manufacturing Technology Institute, Bengaluru – 560022, Karnataka, India,

<sup>3</sup>Department of Materials Engineering, Indian Institute of Science, Bengaluru, Karnataka, India

<sup>4</sup>Environmental Science and Industrial Development Department, Faculty of Postgraduate Studies for Advanced Sciences, Beni-Suef University, 62511, Beni-Suef, Egypt

<sup>5</sup>Interdisciplinary Research Centre for Membranes and Water Security, King Fahd University of Petroleum and Minerals, Dhahran, 31261, Saudi Arabia

<sup>6</sup>Department of Botany, Nagaland University, Lumami, 798627, Nagaland, India

Corresponding author: <a href="mailto:onegroupb203@gmail.com">onegroupb203@gmail.com</a>



Scheme S1: A schematic flow diagram for the formation of ZIF-67



Scheme S2: Synthesis procedure for ZIF-67/PVA nanofibers

Table S1: Systems utilized for molecular docking

System	Macromolecule	Ligand
ZIF67-PVA	ZIF67	PVA
ZIF67-Pb	ZIF67	Pb
PVA-Pb	PVA	Pb
ZIF67-PVA-Pb	ZIF67-PVA	Pb

System	Topmost Docked Structure*	Binding Energy (Kcal mol <sup>-1</sup> )	Type of Interactions
ZIF67-PVA		+1.85	Hydrogen bonding (C-H)
ZIF67-Pb		-2.07	Electrostatic and hydrogen bonding (C-H)
PVA-Pb	venter	-2.90	Hydrogen bonding (conventional and C-H)
ZIF67-PVA- Pb		-3.30	Electrostatic and hydrogen bonding (conventional and C-H)

 Table S2. Results obtained using molecular docking simulations for predicting interactions in various systems

\* Orange: Electrostatic interaction, Light green: C-H Hydrogen bond, Yellow: Conventional Hydrogen bond

Model	Co	10.0	20.0	30.0	40.0	50.0
Pseudo 1 <sup>st</sup>	$K_1$	0.030	0.022	0.016	0.004	0.001
order model	$q_e$	42.4	63.3	99.4	283.5	1808.7
	$R^2$	0.98	0.98	0.98	0.97	0.95
Pseudo 2 <sup>nd</sup>	$K_2 [g mg^{-1} min^{-1}]$	0.0005	0.0002	9.23E-05	0.0001	1.79E-05
order model	$q_e$	53.2	84.4	139.6	139.6	303.3
	$R^2$	0.99	0.99	0.98	0.89	0.89
The Mixed 1,	K	0.001	0.0006	0.001	0.0005	4.83E-06
2-order model	$q_e$	52.1	82.8	136.0	446.4	23.39 E4
	$f_2$	0.961	0.966	0.920	0.805	2.7E-07
	$R^2$	0.99	0.99	0.99	0.97	0.95
Avrami model	$q_e$	42.4	63.3	99.4	284.4	6828.8
	$k_{av}$	0.179	0.154	0.129	0.067	0.013
	n <sub>av</sub>	0.169	0.145	0.123	0.064	0.013
	$R^2$	0.98	0.98	0.98	0.97	0.95
Intraparticle	$k_{ip}$	3.88	5.59	7.75	9.20	10.11
diffusion	C <sub>ip</sub>	2.7	0.8	0	0	0
model	$R^{2}$ [-]	0.98	0.99	0.99	0.91	0.79

 Table S3: Calculations of the different kinetic models for Pb (II) sorption onto ZIF 

 67/PVA nanofibers

Where  $C_0$  in mg/L,  $K_1$  in min<sup>-1</sup>,  $q_e$  in mg g<sup>-1</sup>,  $K_2$  in g mg <sup>-1</sup> min <sup>-1</sup>,  $k_{ip}$  in mg/g.min<sup>1/2</sup> and

 $c_{ip}$  in mg g<sup>-1</sup>

Table S4: Parameters of the isotherm adsorption for the Pb<sup>2+</sup> sorption to ZIF-67/PVA nanofibers

Model	Parameter	Value	Parameter	Value
Langmuir (linear)	$q_{max} [mg. g^{-1}]$	109.9	<b>R</b> <sup>2</sup>	0.81
	b	0.37		
Langmuir (non-linear)	$q_{max}$ [mg. g <sup>-1</sup> ]	364.49	R <sup>2</sup>	0.86
	K <sub>L</sub>	0.027		
Freundlich (linear)	n	2.297	<b>R</b> <sup>2</sup>	0.88
	K <sub>F</sub> [Lmg <sup>-1</sup> ]	30.58		
Freundlich (non-linear)	n	1.56	<b>R</b> <sup>2</sup>	0.91
	K <sub>F</sub> [Lmg <sup>-1</sup> ]	18.35		
Dubinin-Radushkevich	$q_{m}[mg. g^{-1}]$	226.73	$\mathbb{R}^2$	0.99
	Kad	0.0024		
Khan	$q_{m}[mg. g^{-1}]$	0.208	a <sub>K</sub>	0.361
	b <sub>K</sub>	1101	$R^2$	0.91
Fritz-Schlunder	$q_{mFSS} [mg. g^{-1}]$	31.21	m <sub>1</sub>	0.639
	K1	0.740	m <sub>2</sub>	0
	K <sub>2</sub>	0.259	$\mathbf{R}^2$	0.91



Figure S1. Effect of different parameters on Pb<sup>2+</sup> adsorption (a) Variation of pH: 2-7 ( ZIF-67/PVA nanofibers 20 mg/L, Pb<sup>2+</sup> concentration 10 mg/L, time of contact 2 h), (b) Point of zero charge (c): Variation of ZIF-67/PVA nanofibers dosage (pH 6, ZIF-67/PVA nanofibers 20-100 mg/L, Pb<sup>2+</sup> concentration 10 mg/L and time 2 h) & (d) Effect of Pb concentration (pH 6, ZIF-67/PVA nanofibers 20 mg/L, Pb<sup>2+</sup> concentration 10 - 50 mg/L and time of contact 2 h).



Figure S2. Fitting of the experimental dats to the PFO, PSO, IPD, mixed 1<sup>st</sup> and 2<sup>nd</sup> order, and Avrami models for Pb<sup>2+</sup> adsorption onto ZIF-67/PVA nanofibers at different initial concentrations: (a) 10, (b) 20 (c) 30 (d) 40 and (e) 50 mg L<sup>-1</sup>.



Figure S3. Fitting the experimental data of the Pb<sup>2+</sup> adsorption onto ZIF-67/PVA nanofibers using: (a) Langmuir (linear), (b) Freundlich (linear), (c) Langmuir (non-linear), (d) Freundlich (non-linear), (e) Dubinin–Radushkevich, (f) Khan, and (g) Fritz-Schlunder models.



Figure S4. (a) XPS spectra of C 1s in zif-67, (b) N 1s spectrum in zif-67, (c) O 1s spectrum in zif-67, (d): XPs of C after electrospinning and (e) high-resolution spectra of C 1s after Pb (II) adsorption.



Figure S5: Cyclic stability of ZIF-67/PVA nanofibers after seven cycles