

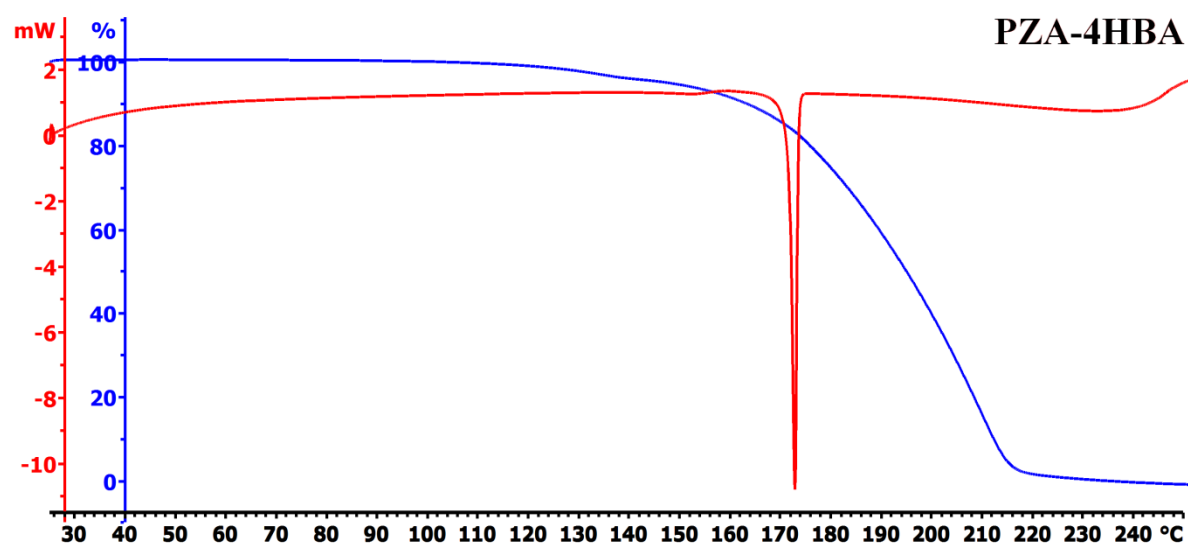
## Preparation of Pyrazinamide Eutectics versus Cocrystals Based on Supramolecular Synthons Variations

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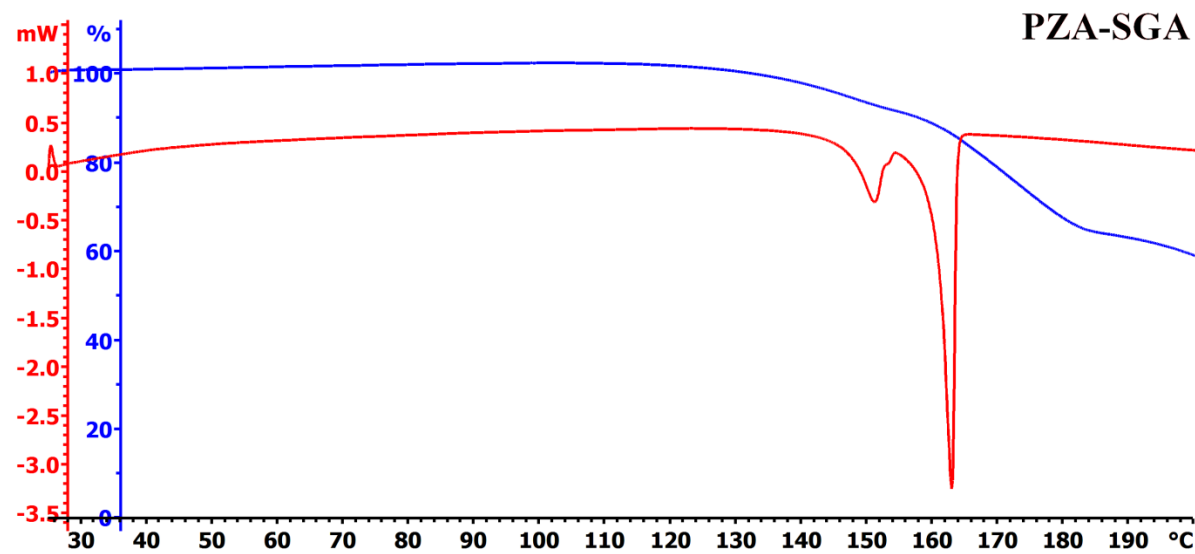
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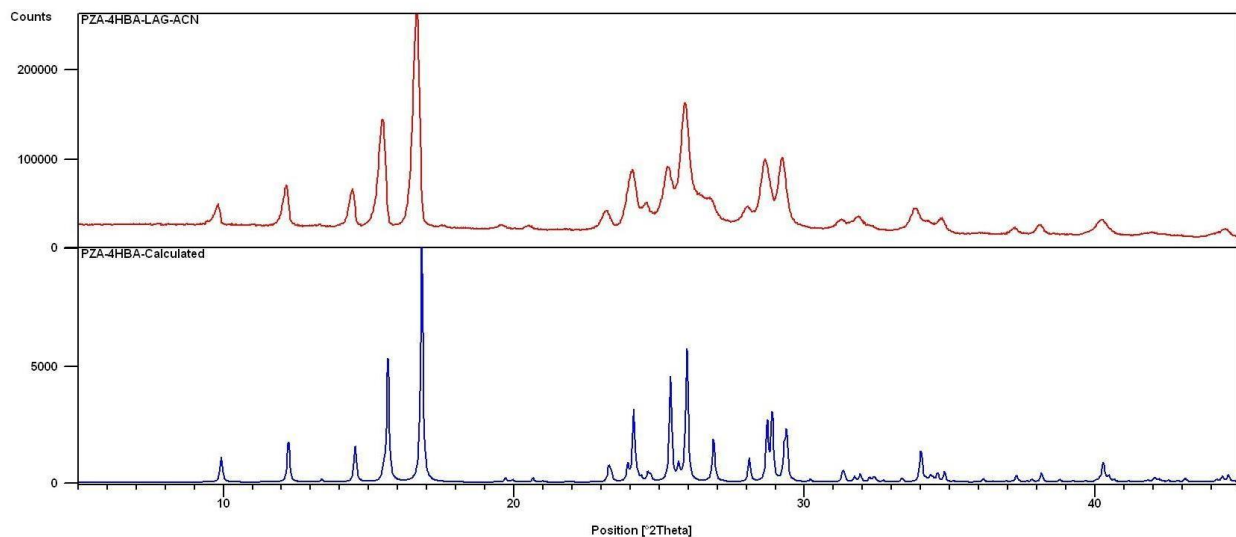


(a)

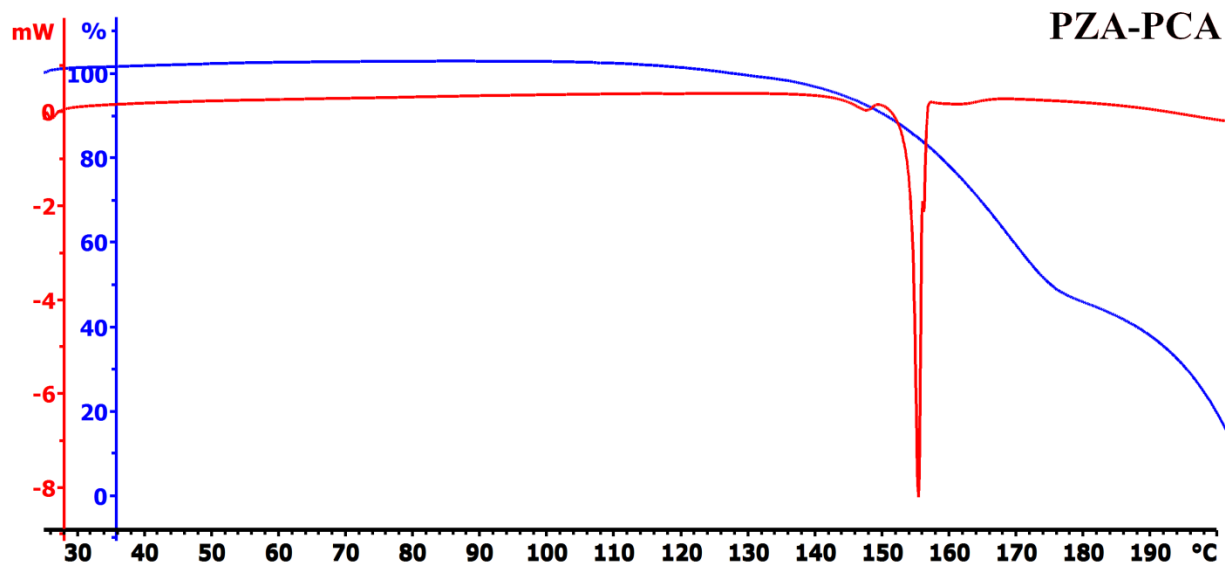


(b)

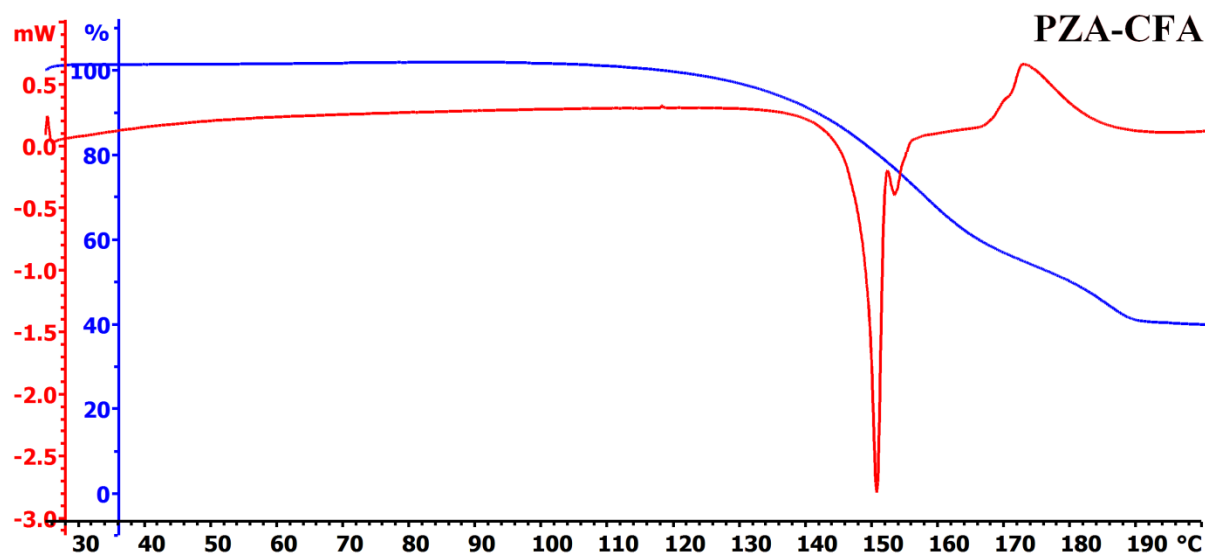
**Figure S1.** DSC (red) and TGA (blue) thermograms of (a) PZA–4HBA and (b) PZA–SGA cocrystals show no phase transition and corresponding weight loss pertaining to solvent incorporation before the melting endotherm indicating that the combinations are non-solvated cocrystals. The cocrystals (PZA–4HBA 172 °C, PZA–SGA 162 °C) have melting point less than their respective parent components (PZA 190 °C, 4HBA 213 °C, SGA 205 °C).



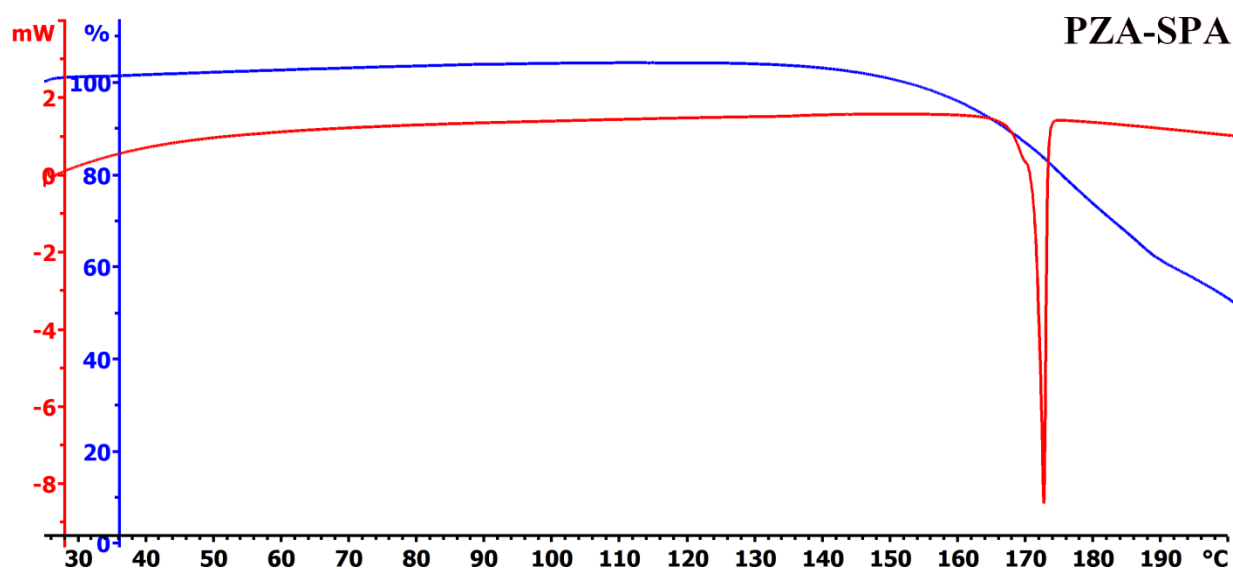
**Figure S2.** Experimental PXRD pattern (top) of PZA–4HBA cocrystal prepared using LAG shows complete match with that of the calculated powder pattern (bottom) obtained from the X-ray crystal structure.



(a)

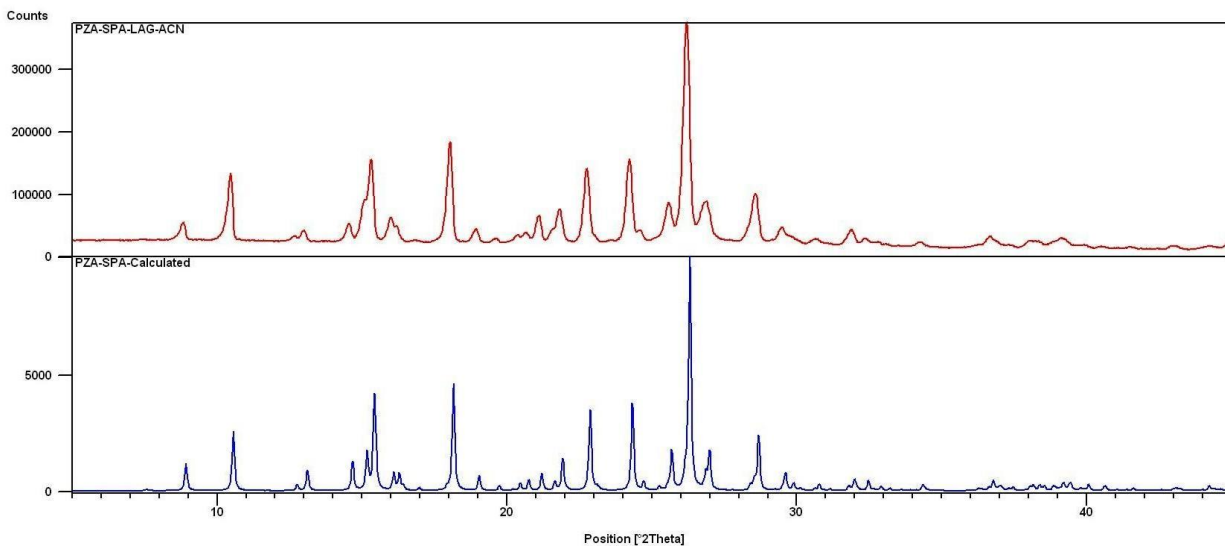


(b)

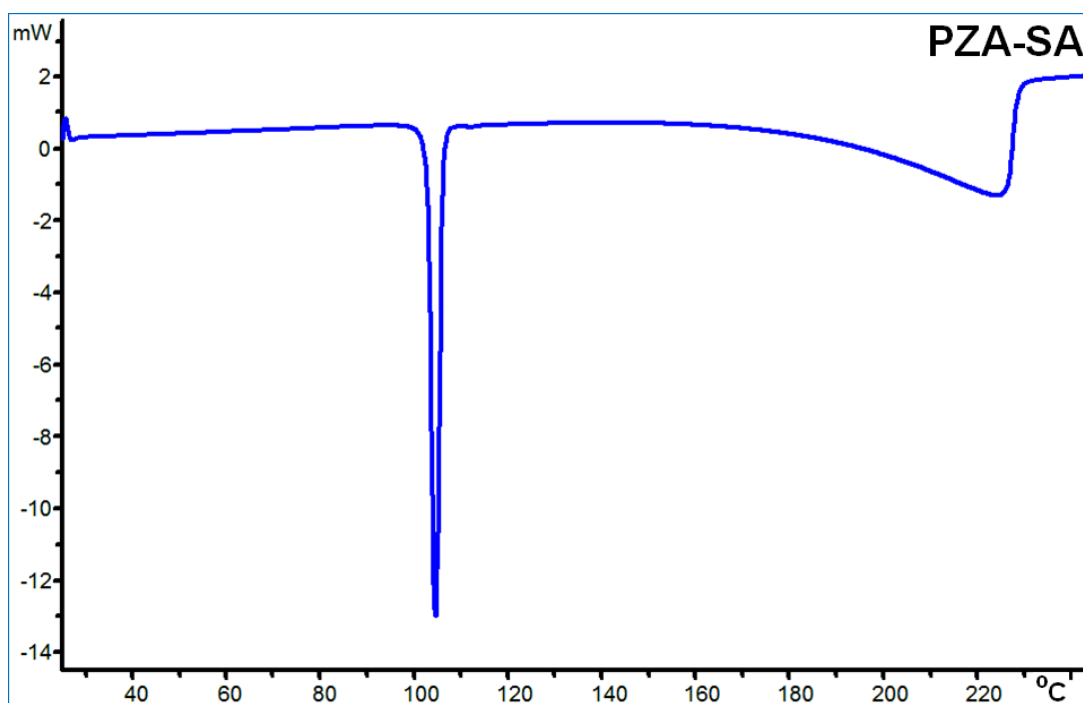


(c)

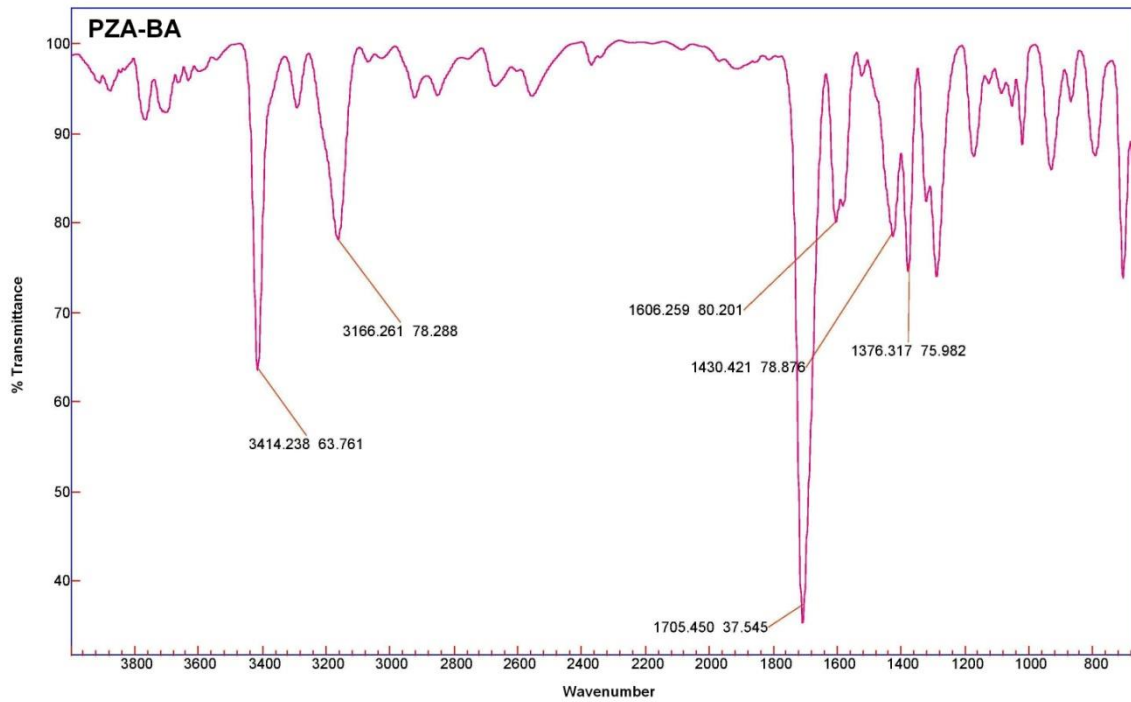
**Figure S3.** DSC (red) and TGA (blue) thermograms of (a) PZA–PCA, (b) PZA–CFA and (c) PZA–SPA cocrystals show no phase transition and corresponding weight loss pertaining to solvent incorporation before the melting endotherm indicating that the combinations are non-solvated cocrystals. The cocrystals (PZA–PCA 154 °C, PZA–CFA 150 °C, PZA–SPA 172 °C) have melting point less than their respective parent components (PZA 190 °C, PCA 214 °C, CFA 211 °C, SPA 202 °C).



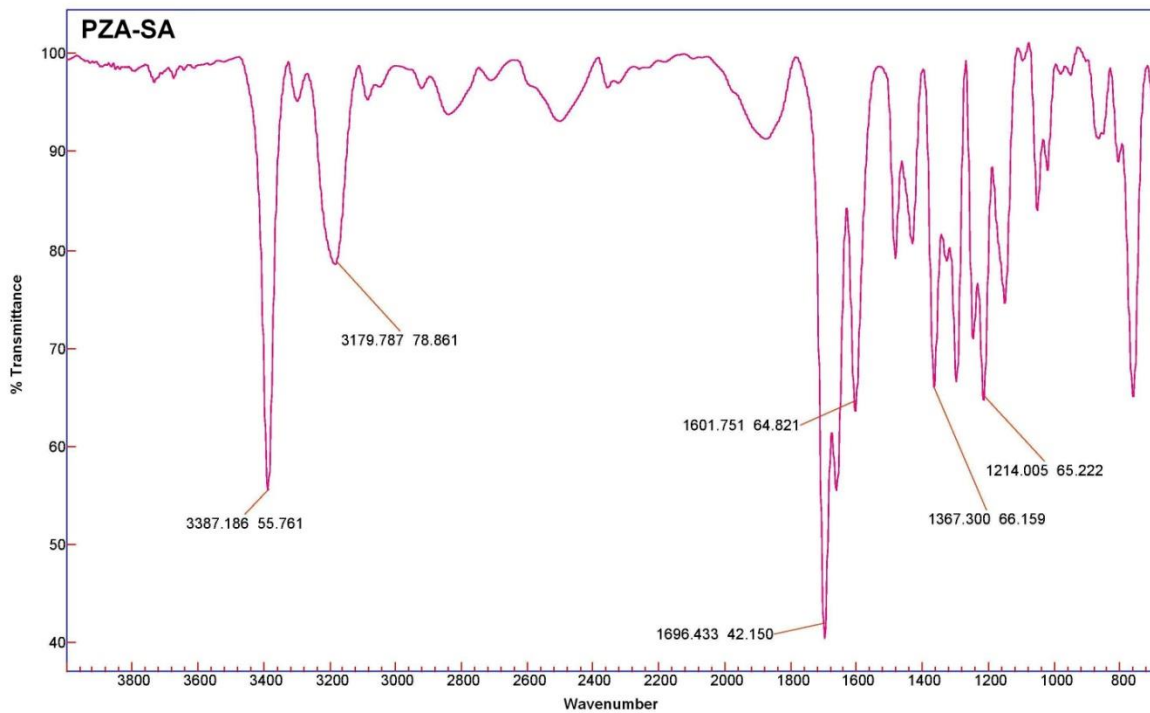
**Figure S4.** Experimental PXRD pattern (top) of PZA–SPA cocrystal prepared using LAG shows complete match with that of the calculated powder pattern (bottom) obtained from the X-ray crystal structure.



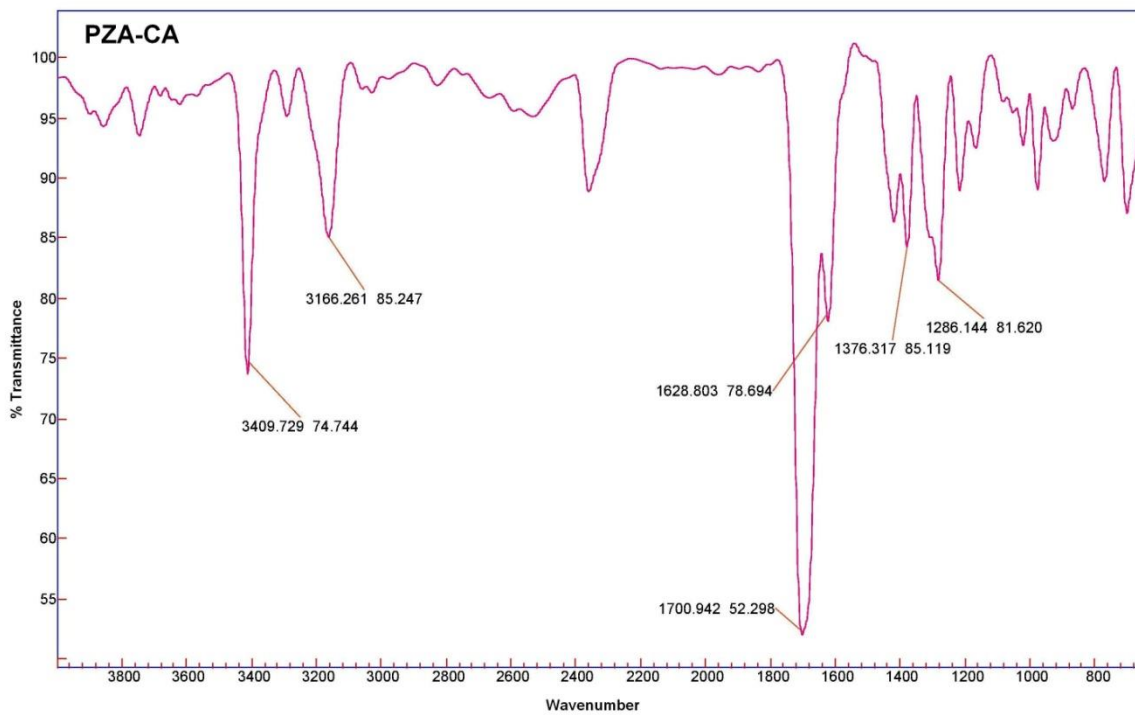
**Figure S5.** DSC thermograms of PZA-SA eutectic at its eutectic compositions i.e. 1:1 showing lowest melting endotherm coincides with the binary phase diagram.



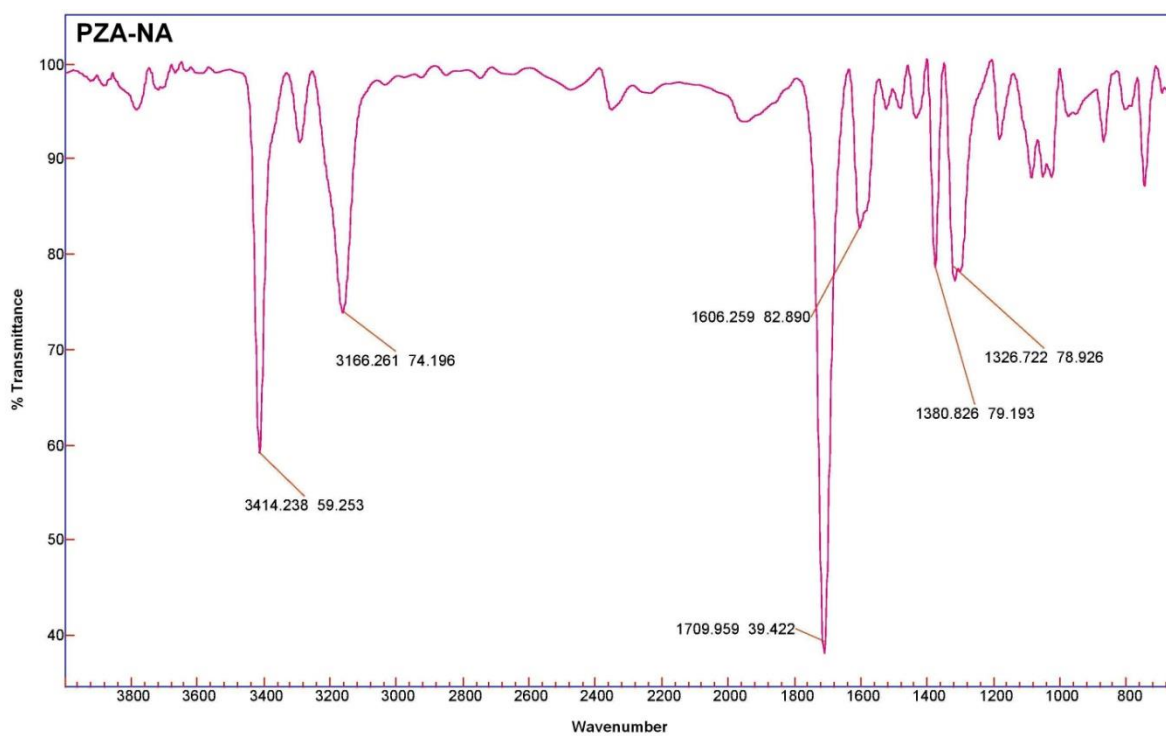
(a)



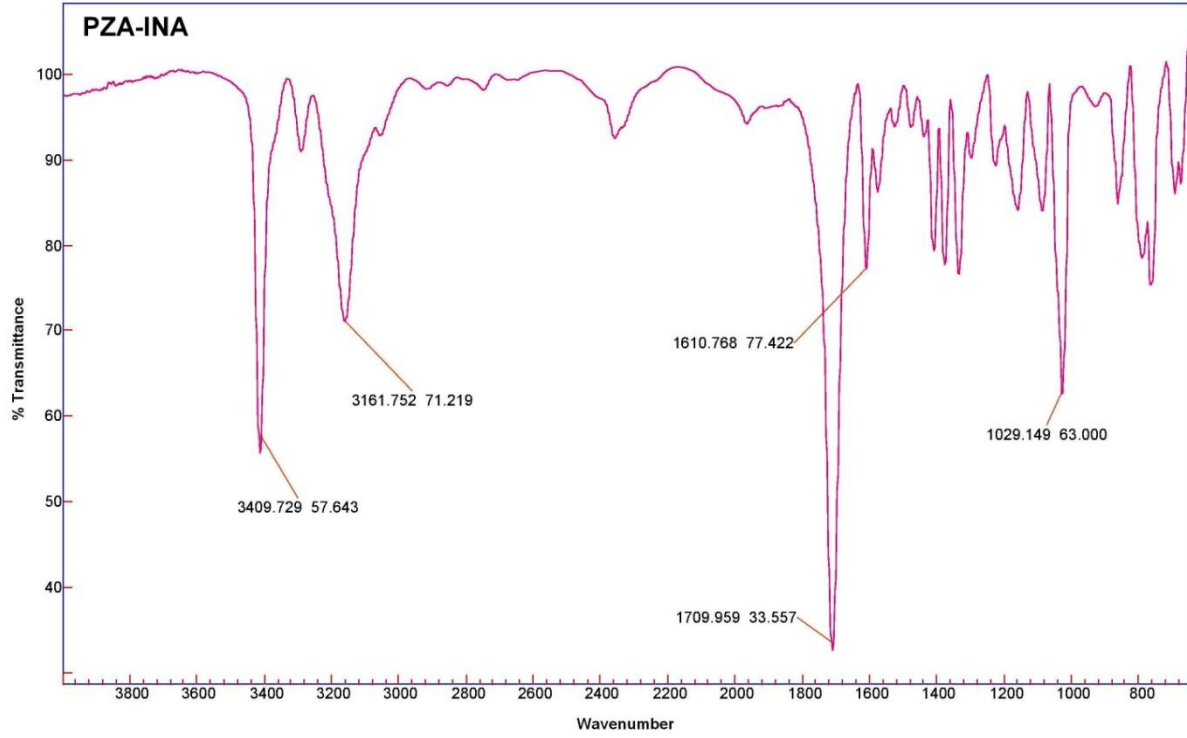
(b)



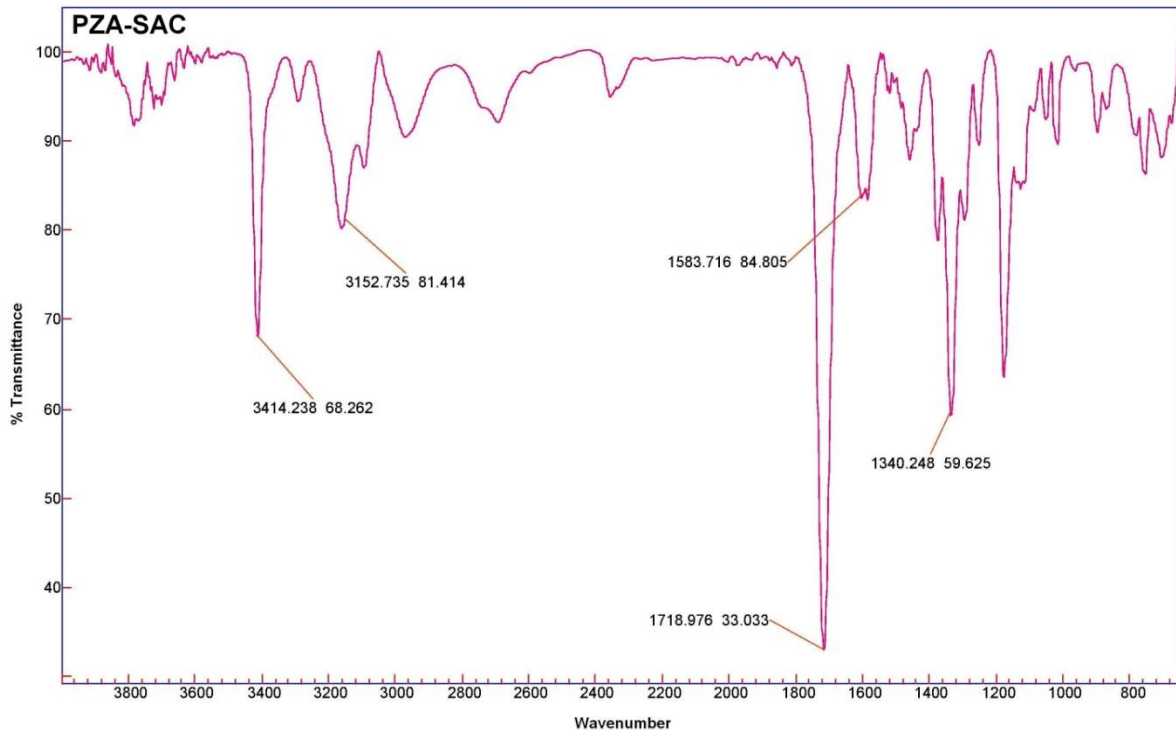
(c)



(d)



(e)



(f)

**Figure S6.** FT-IR spectra of the respective eutectic systems (a) PZA-BA, (b) PZA-SA, (c) PZA-CA, (d) PZA-NA, (e) PZA-INA and (f) PZA-SAC.